Magnetic Interactions in $CoM_x^{3+}Ga_{2-x}O_4$ Spinel Solid Solutions: II. $CoAl_xGa_{2-x}O_4$

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Magnetic measurements have been performed in the range 4.2 to 300° K on the spinel solid solutions $CoAl_xGa_{2-x}O_4$ with $0.0 \le x \le 2.0$. The magnetic properties vary with cobalt ion distribution among T_d and O_h sites; the influence of trivalent diamagnetic ions in determining the relative strengths of the various exchange interactions is discussed.

Introduction

In our previous paper on $\operatorname{CoRh}_x \operatorname{Ga}_{2-x} O_4(1)$ we correlated the observed magnetic behavior with cobalt ions distribution between A and B sites of spinel, obtaining information about the relative strength of the various exchange interactions.

In the $CoAl_xGa_{2-x}O_4$ solid solution, the Co^{2+} site distribution varies with composition x (2), exhibiting the same trend as was observed for the previous system, i.e., the Co^{2+} octahedral occupation increases with the gallium content, giving rise to concomitant AA, BB, and AB interactions.

The antiferromagnetic spinels $CoRh_2O_4$ ($T_N = 25^{\circ}K$) and $CoAl_2O_4$ ($T_N = 5^{\circ}K$) have been studied extensively, and the higher Néel temperature for $CoRh_2O_4$ supported the suggestion that the trivalent ion is involved in the indirect exchange coupling between Co^{2+} ions in A sites (3, 4). In this work we analyze the dependence of the magnetic properties on the site distribution of the Co^{2+} ions in order to obtain evidence for the different role of Rh³⁺ compared to Al³⁺ in the exchange interactions.

Experimental Procedures

Sample preparation and cation distribution are reported elsewhere (2). Magnetic measurements were performed for the samples prepared at 1073°K on a Faraday-type balance described previously (1). The experimental susceptibility, obtained in the range 4.2 to 300°K at H = 10 kOe, was corrected for diamagnetism using the table of Selwood (5).

Experimental results

Magnetic data and cation distribution are reported in Table I. The Curie constant C and the asymptotic temperature θ were determined by fitting the linear parts of χ^{-1} vs T plots with a least-squares program.

A survey of susceptibility measurements is given in Figs. 1 and 2. A careful investigation of the temperature range 4.2 to 10°K has shown the existence of a Néel point for all the samples, pointing to the presence of long-range antiferromagnetic ordering.

Below $\sim 80^{\circ}$ K, all the samples except CoAl₂O₄ showed a large deviation from

	TA	BL	Æ	Ι
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x	$\operatorname{Co}(O_h)^b$	$T_{N}(^{\circ}K) \pm 1$	T.I.P. ^c (emu × 10 ⁶)	$-\theta$ (°K) ± 1		$C \pm 0.02$		$\mu_{\rm eff}$ (B.M.) \pm 0.02	
				а	b	а	b	а	b
2.00	18	5	460	97	113	2.14	2.40	4.14	4.38
1.25	36	8	359	70	. 83	2.43	2.65	4.41	4.61
1.00	49	8	286	65	70	2.56	2.69	4.53	4.64
0.50	67	8	185	55	58	2.61	2.70	4.57	4.65
0.25	72	7	157	60	63	2.84	2.91	4.77	4.83
0.00	74	10	146	65	69	3.01	3.10	4.91	4.98

MAGNETIC DATA FOR COAl, Ga1, O4 SOLID SOLUTIONSª

^a a, with T.I.P. correction; b, without T.I.P.

^b From Ref. (2).

^c Calculated (see text). Only the value for x = 2.00 is taken from experiments (see 6).

the Curie–Weiss law, attributed, as in $CoRh_xGa_{2-x}O_4$ (1), to the presence of small uncompensated clusters and to isolated paramagnetic Co^{2+} ions.

This suggestion is supported by the M vs H curves (Fig. 3) obtained at 5°K up to a maximum field of 70 kOe. Indeed, for CoAl₂O₄ a linear plot was observed, but for



FIG. 1. Reciprocal magnetic susceptibility as a function of temperature with an applied field H = 10 kOe.

other specimens there was a Brillouin-type behavior, attributed to paramagnetic ions and uncompensated clusters.

Composition Dependence of C and θ

Given the lack of data in the hightemperature range it was not possible to introduce an appropriate correction for temperature-independent paramagnetism. Nevertheless, the T.I.P. term for Co^{2+} ions in



FIG. 2. Reciprocal magnetic susceptibility as a function of temperature with an applied field H = 10 kOe.



FIG. 3. Molar magnetization at 5° K as a function of applied field.

 T_d sites is not negligible, as found by Cossee (6) for CoAl₂O₄ (T.I.P. = 460 × 10⁶ emu).

To estimate the influence of T.I.P. correction on the values of C and θ we have evaluated the T.I.P. term (Table I) for each composition as a function of the tetrahedral occupation.

We used the value quoted by Cossee for $CoAl_2O_4$, assuming that his sample had the same cation distribution as ours (i.e., 82% of cobalt ions in T_d sites).

We remark that this is a rough estimate and that we do not expect a behavior linear with the composition of the T.I.P. value for Co^{2+} ions in T_d sites because of the gradual change in lattice site symmetry and crystal field parameters (T.I.P. = $8N\beta^2/10 Dq$ for the ${}^{4}A_{2}$ term). Indeed, the experimental T.I.P. values for the CoRh_xGa_{2-x}O₄ series (1) do not change linearly with the Rh³⁺ content and tetrahedral Co²⁺ ion occupation.

The comparison between the values of C and θ evaluated with T.I.P. correction and without shows that the absolute values are affected by this correction. Therefore, the C and θ values reported (Table I) should be considered only apparent, and the accuracy reported refers only to the precision limits of the instrument. However, the behavior of C and θ (Figs. 4 and 5) remains substantially the same in both cases.



FIG. 4. Variation of the Curie constant C with percentage $Co(O_h)$. Full line, C values calculated without T.I.P. correction (\bigcirc); dashed line; with T.I.P. correction (\bigcirc).



FIG. 5. Variation of T_N ($\textcircled{\bullet}$) with percentage Co(O_h). Variation of θ (\blacksquare , \Box) with percentage Co(O_h). Full line, θ values calculated without T.I.P. correction; dashed line, with T.I.P. correction.

The nonlinear behavior of C with percentage Co^{2+} in O_h sites (Fig. 4) is due to the change of C_A and C_B within $CoAl_xGa_{2-x}O_4$ solid solutions, probably caused by the presence of lattice distortions. The negative value of θ obtained from the Curie-Weiss law confirms the predominance of antiferromagnetic interactions for all the samples. There is (Fig. 5) a decrease in θ up to 67% of Co^{2+} in B sites, and then an increase.

Discussion

The results confirm that the distribution of Co^{2+} among the A and B sites determines the

magnetic behavior of spinel solid solutions. Comparing these results with that obtained for the $\operatorname{CoRh}_x \operatorname{Ga}_{2-x} O_4$ system (1) leads to the following conclusions.

 $CoRh_2O_4$ and $CoAl_2O_4$ are both antiferromagnetic. While the former, which is a normal spinel, substantially follows the Néel twosublattice theory, the latter, being partially inverse, exhibits a high $|\theta/T_N|$ ratio ($\simeq 18$), revealing the presence of more than one type of interaction. Therefore, a comparison of T_N between $CoAl_2O_4$ and $CoRh_2O_4$ is not clearly diagnostic of the role of the trivalent ion in determining the strength of AA interactions. Indeed, as we have previously shown (1), the AB interactions tend to destroy the AA longrange ordering.

In the plot of T_N vs percentage $Co(O_n)$ (Fig. 5) the initial increase of T_N with increase in the number of AB interactions suggests that these interactions stabilize a long-range order. This order is not simple and would take the form of an antiferromagnetic interaction between small clusters.

The higher T_N value for CoGa₂O₄ than for the other samples could be due to a new kind of antiferromagnetic ordering, resulting from the new balancing between the different exchange interactions.

The behavior of θ with Co²⁺ site distribution (Fig. 5) is quite different from that observed for the CoRh_xGa_{2-x}O₄ system. The progressive decrease in $|\theta|$ with increase in cobalt octahedral occupation supports the presence of ferromagnetic interactions.

These interactions may be due to cobalt ions in B sites, which would be ferromagnetically coupled by the strong ABinteractions, as found elsewhere (7).

The subsequent increase in $|\theta|$ above 67% of octahedral cobalt ions reveals an increase in antiferromagnetic interactions, which may be due to the increasing importance of BB next-nearest-neighbor interactions.

Conclusions

The magnetic properties of both spinel solid solutions $CoAl_xGa_{2-x}O_4$ and $CoRh_xGa_{2-x}O_4$ depend on the Co^{2+} distribution between O_h and T_d sites. This distribution determines the occurrence of spin uncompensated clusters and isolated ions, as revealed by lowtemperature magnetic behavior.

The observed dependence of T_N and θ on cation distribution is different for the two systems and this fact indicates that the presence of Al^{3+} instead of Rh^{3+} influences the relative strength and the balancing between the various exchange interactions *AA*, *BB*, and *AB*.

Our results support the view that AA interactions are more effective when $M^{3+} = Rh^{3+}$, confirming the intermediate role of Rh^{3+} , as proposed by Blasse (4). Indeed, in the spinel solid solutions containing Rh^{3+} ions it appears that antiferromagnetic order is established between spins in A sites, and that AB interactions tend to destroy it. Conversely, when Rh^{3+} is replaced by Al^{3+} ion, the AA interactions become less important, and the magnetic order seems to be stabilized by the AB interactions.

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